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Holt Chemistry 6 Stoichiometry Section: Limiting Reactants and Percentage Yield ... Answer the following items in the space provided. 11. ... Holt Chemistry 86 Stoichiometry Answer Key TEACHER RESOURCE PAGE. 12. 23 g C 2H 5OH (1 mol C 2H 5OH/46.08 g C 2H 5OH) 0.50 mol C 2H 5OH 32 g O

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Holt ChemFile: Problem-Solving Workbook 57 Mole Concept Name Class Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is about 1/5 of the molar mass of boron. Therefore, 2.00 g boron will contain about 1/5 of an Avogadro's constant of atoms. Practice 1. Calculate the number of atoms in each of the following masses:

Skills Worksheet Problem Solving - Penn Arts & Sciences

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