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PEARSON CHEMISTRY: The law of conservation of mass is explained in Lesson 2.4 (p. 50). Students are introduced to balancing chemical equations in Lesson 11.1 (pp. 346- 354). Students learn to write and balance the five different types of chemical reactions in Lesson 11.2 (pp. 356-367).

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an element have the same chemical properties. Radioactive isotopes have unstable nuclei and break down at a constant rate. Chemical Compounds A chemical compound is a substance formed by the chemical combination of two or more elements in definite proportions. The physical and chemical

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84 Guided Reading and Study Workbook SECTION 9.4 NAMING AND WRITING FORMULAS FOR ACIDS AND BASES (pages 271-273) This section explains the three rules for naming acids and shows how these rules can also be used to write the formulas for acids.

Name Date Class CHEMICAL NAMES AND FORMULAS 9

Representative Particle Chemical Formula Representative Particles in 1.00 mol Atomic oxygen O Oxygen gas molecule Sodium ion Sodium chloride false Use the chemical formula to find the number of atoms in one molecule and multiply this number by Avogadro's number, the number of particles in one mole. atom 6.02 • 10²³ O 2 6.02 • 10²³ ion ...

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