

Physical Chemistry Atkins Solution Manual



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"Microprojects with Mathcad" can be used as a supplement in university courses that use many of the currently available physical chemistry lecture and laboratory texts, especially the popular "Physical Chemistry" by Peter Atkins.

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At the normal boiling temperature, the partial vapour pressures must add up to 1 atm (101.3 kPa). These ideal partial vapour pressures add up to only 90.7 kPa, so the solution is not ideal (b) We

actually do not have enough information to compute the initial composition of the vapour above the solution.

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